

Solution, what would be the density of Hg = 13600 kgm^{-3}).

8. The radii of the bigger piston and the smaller piston are in the ratio 50:4. If 20N acts on the smaller piston, calculate the force on the bigger piston.
9. The normal pressure of air is 76 cm of Hg. Calculate the pressure in SI units. (Take $g = 10 \text{ ms}^{-2}$, and density of Hg = 13600 kgm^{-3}).

10. Calculate the pressure exerted by 0.12 m vertical length of alcohol of density 0.12 gcm $^{-3}$ in SI units. (Take $g = 10 \text{ ms}^{-2}$)

contd.